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1. The percent
composition of a
compound is the
percent by mass of
each of the elements

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in a compound. 2.

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(160 flashcards) The

sum of the atomic

masses of all the...

The number of atoms

in one mole of an e...

Which compound has

a molar mass of

174....

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mass. the molar mass
of an element

expressed in grams
per mole (g/mol)

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formula units in a
ionic compound bond.
Mole. the simplest
structural unit of an
element or
compound.

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STUDY. PLAY. Mole.

SI base unit used to
measure the amount
of a substance and
communicates how
much of a substance
is present. Avogadro's
number. 6.022×10^{23}
molecules. Molar
mass. Mass in grams
of one mole of any

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pure substance, has
the units g/mol.

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Reactant: The

uranium(IV) oxide,

UO₂, which is used as

fuel in nuclear power

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plants, has a higher percentage of the fissionable isotope uranium-235 than is present in the UO_2 found in nature. To make fuel-grade UO_2 , chemists first convert uranium oxides to uranium hexafluoride, UF_6 , whose concentration of uranium-235 can ...

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