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10 163 CHAPTER 10 SOLUTIONS

MANUAL 3. Apply How many atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two

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significant digits. 1.007 g  
1 H-1 atom  $\times$   $1.672 \times 10^{-24}$  g  
a.  $6.023 \times 10^{23}$  H-1 atoms 4.

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During evaporation,  
particles in a liquid gain  
enough

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pages 282-288 Practice 16

Problems pages 284-287 Write  
skeleton equations for the  
following word equations. 1.  
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react to yield hydrogen  
bromide.

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